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Chemistry Chapter 8: Thermochemistry
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Instructure

Sodium chloride: It is ionic in nature so it dissolves in water. Its ions and can move easily in molten solution and hence conducts electricity. Therefore, it can act as an electrolyte. The presence of chloride ion can be confirmed by the following test. When silver nitrate solution is added to sodium chloride

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solution a white thick precipitate of silver chloride is formed, this indicates the ...

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